Photorearrangements of 1-Cyanocyclohexenes. A "Type A" Process Involving the Acrylonitrile Chromophore

By CARL MANNING and JOHN J. McCullough*

(Chemistry Department, McMaster University, Hamilton, Ontario L8S 4M1, Canada)

Summary Irradiation of simple 1-cyanocyclohexenes gives derivatives of cyanobicyclo[3.1.0]hexanes, for which structural evidence is presented.

The di- π -methane rearrangement¹ has been extensively studied from the viewpoints of structural dependence, stereochemistry, and mechanism. In contrast, photorearrangements of the σ molecular framework have been less widely studied, the best known case being the 'type A' arrangement (e.g. $1 \rightarrow 2$) of α,β -unsaturated ketones.² The



latter reaction has been the subject of kinetic³ and stereo-chemical⁴ studies.

We now report a photorearrangement of 1-cyanocyclohexene which formally resembles the 'type A' rearrangement³ of enones.

Irradiation of 1-cyanocyclohexene (3), in purified hexane, with the full arc of a Hanovia type L 450W lamp in a watercooled quartz immersion well gave 1-cyanobicyclo[3.1.0]hexane (4), and the *exo* and *endo* isomers of 6-cyanobicyclo-[3.1.0]hexane (5). The products (4) and (5) were formed in equal amounts.

These products were isolated by preparative $g.l.c.,\dagger$ and their structures were proved by synthesis. Product (4) was

prepared from 1-cyanocyclopentane by treatment with dimethylsulphoxonium methylide⁵ in dimethyl sulphoxide.



Reaction of cyclopentene with cyanocarbene⁶ (CNCH:) gave the epimers of (5). The synthetic materials and photolysis products had identical n.m.r. spectra, and g.l.c. behaviour. Both the compounds were hydrogenolysed, using 10% Pd/C in 95% ethanol at 25 lb in⁻² and room temperature, to afford cyanocyclohexane.

Irradiation of 1-cyano-3,3,5,5-tetramethylcyclohexene (6), which can be synthesized readily from isophorone, gave products identified as the bicyclohexanes (7) and (8) in a



ratio of 80:20, respectively. These compounds were isolated and purified by column chromatography and g.l.c. The i.r. spectra showed an unconjugated nitrile band at 2230 cm⁻¹. The structures were assigned from the 220

 \dagger Preparative g.l.c. was performed on 5' \times $\frac{1}{2}$ " of 7% QF-1 on 60-70 mesh Chromosorb W with helium at 90 ml/min. (4) and (5) were resolved at 80 °C, and (7) and (8) at 120 °C.

MHz proton n.m.r. spectra. The spectrum of (8) is simple owing to the symmetry of the molecule: $\delta 0.87$ and 1.28(each 2H, d, J 14.0 Hz, fing CH2), 1.01 [1H, t, J 3.5 Hz, >C(CN)H, partially obscured by Me signal], 1.02 and 1.21 (each 6H, s, Me), and 1.77 (2H, α , bridgehead CH).

The ¹H n.m.r. spectrum of (7) showed resonances at δ 0.98, 1.06, 1.17, and 1.31 (all 3H, s, Me), 1.18 (1H, dd, Jab 7, $J_{b,c}$ 4.0 Hz, H^b, partially obscured by Me signal), 1.51 (1H, d, J_{d.e} 14 Hz, H^e), 1.63 (1H, m, H^a), 1.77 (1H, dd, J_{a,e}

2, $J_{d,e}$ 14 Hz), and 1.93 (1H, dd, $J_{a,c}$ 13.0, $J_{b,c}$ 4.0 Hz, H^c). The value of 2 Hz for $J_{a,e}$ is reasonable for W coupling⁷ between H^a and H^e. The ¹³C-n.m.r. spectrum was also consistent with the assigned structure.

We thank Dr. A. Grey for the 220 MHz spectra, and the National Research Council of Canada for financial support.

(Received, 25th October 1976; Com. 1198.)

¹ A comprehensive review is available: S. S. Hixon, P. S. Mariano, and H. E. Zimmerman, *Chem. Rev.*, 1973, 73, 531. ² O. L. Chapman, T. A. Rettig, A. I. Dutton, and P. Fitton, *Tetrahedron Letters*, 1963, 2049.

³ H. E. Zimmerman, R. G. Lewis, J. J. McCullough, A. Padwa, S. W. Staley, and M. Semmelhack, J. Amer. Chem. Soc., 1966, 88, 1965.

⁶D. I. Schuster and R. H. Brown, J. C. S. Chem. Comm., 1976, 28, and refs. therein.
⁶E. J. Corey and M. Chaykovsky, J. Amer. Chem. Soc., 1965, 87, 1353.
⁶M. J. S. Dewar and R. Petit, J. Chem. Soc., 1956, 2026. Our procedure was modelled on that of Warkentin for an ethoxycarbonyl-¹Ulticol Mathematical Constraints of Constraints carbene addition: J. Warkentin, E. Singleton, and J. F. Edgar, Canad. J. Chem., 1965, 43, 3456. ⁷L. M. Jackman and S. Sternhell, 'Applications of Nuclear Magnetic Resonance Spectroscopy in Organic Chemistry,' Pergamon,

Oxford, 1969, pp. 324-325.